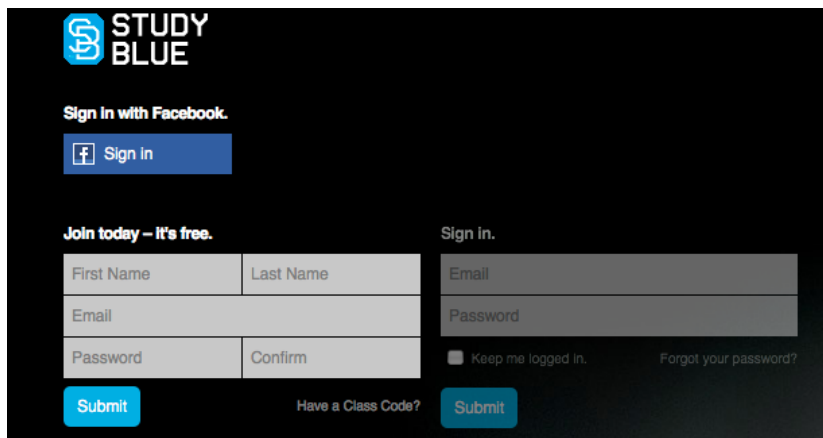


Using StudyBlue

Study Blue allows you to create a free account, then create online flashcards, upload your notes, and share with other students in your class. Since you organize notes by your school and the class name, anyone else at UVM enrolled in the same class will be able to view and use your study materials. Which means you can use their flashcards to study, too. It is a great way to form “study groups” you can participate in at 3:30 in the morning or to share and compare notes for class. I type faster than I print, so it was a quick way to make 200+ vocab flashcards to study for the GRE. And, the site will create quizzes for you to take when it’s time to study. Check it out!

Step One: Create an account.

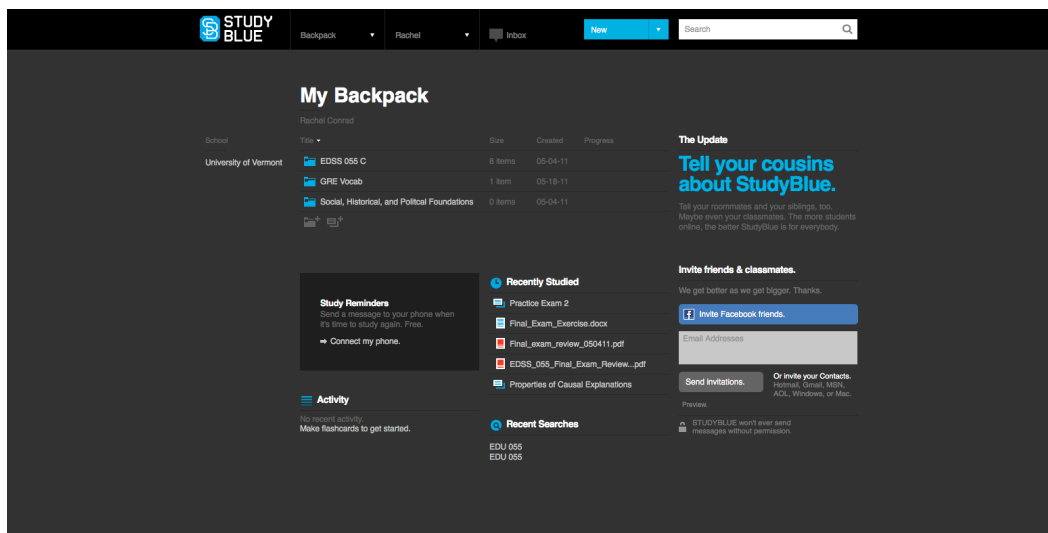
It’s free to join. Just enter your email address and a password and you’re ready to get started.



The image shows the StudyBlue sign-up page. At the top left is the StudyBlue logo. Below it is a 'Sign in with Facebook' button. The main section is titled 'Join today – it's free.' and contains a form with fields for 'First Name', 'Last Name', 'Email', 'Password', and 'Confirm'. There is a 'Submit' button at the bottom left of this form. To the right of the main form is a 'Sign in.' section with fields for 'Email' and 'Password', a 'Keep me logged in.' checkbox, and a 'Forgot your password?' link. At the bottom of the sign-in section is another 'Submit' button. A link 'Have a Class Code?' is located between the two main form sections.

Step Two: Add your UVM courses.

Once you’re logged in, your homepage, or “backpack” will look like this:



The image shows the StudyBlue 'My Backpack' dashboard. At the top is a navigation bar with the StudyBlue logo, 'Backpack', 'Rachet', 'Inbox', and a 'New' button. Below the navigation bar is the 'My Backpack' section. On the left, there is a table of courses for 'University of Vermont' with columns for 'Title', 'Size', 'Created', and 'Progress'. The table lists three courses: 'EDSS 055 C', 'GRE Vocab', and 'Social, Historical, and Political Foundations'. To the right of the table is a 'Study Reminders' section with a 'Connect my phone.' button. Below that is an 'Activity' section. On the right side of the dashboard, there is a 'The Update' section with a 'Tell your cousins about StudyBlue.' link. Below that is a 'Recently Studied' section with a list of items: 'Practice Exam 2', 'Final_Exam_Exercise.docx', 'Final_exam_review_050411.pdf', and 'EDSS_055_Final_Exam_Review...pdf'. At the bottom right is a 'Recent Searches' section with 'EDU 065' and 'EDU 066' listed. There is also an 'Invite friends & classmates.' section with a 'Tell your cousins about StudyBlue.' link and a 'Send Invitations.' button.

When you toggle over the “My Backpack” tab at the top left of the screen, you will be given the option to “Manage Folders.” Select this option.

The screenshot shows the STUDY BLUE interface. At the top, the 'Backpack' tab is selected. The left sidebar shows the 'Manage Folders' option highlighted. The main content area displays a table of folders:

Folder Name	Size	Created	Progress
CHEM 31 Recitation	1 item	Today	0%
EDSS 055 C	8 items	05-04-11	
GRE Vocab	1 item	05-18-11	
Social, Historical, and Political Foundations of Education	0 items	05-04-11	

Below the table, there are sections for 'Study Reminders', 'Recently Studied', and 'Recent Searches'.

The “Manage Folders” screen will look like this:

The screenshot shows the 'Manage Folders' screen. The title 'Manage Folders' is prominently displayed. Below it, there is a section titled 'In my backpack' which lists the folders and their status:

Folder Name	Items	Actions
EDSS 055 C	8 items	Archive. Disconnect.
GRE Vocab	1 item	Archive. Disconnect.
Social, Historical, and Political Foundations	0 items	Archive. Disconnect.

At the bottom, there is a plus icon to add more folders.

Click on the Blue folder without a class title to add a new course to your “backpack.” Enter the name of the course you wish to add. Hit “save” to create the new folder.

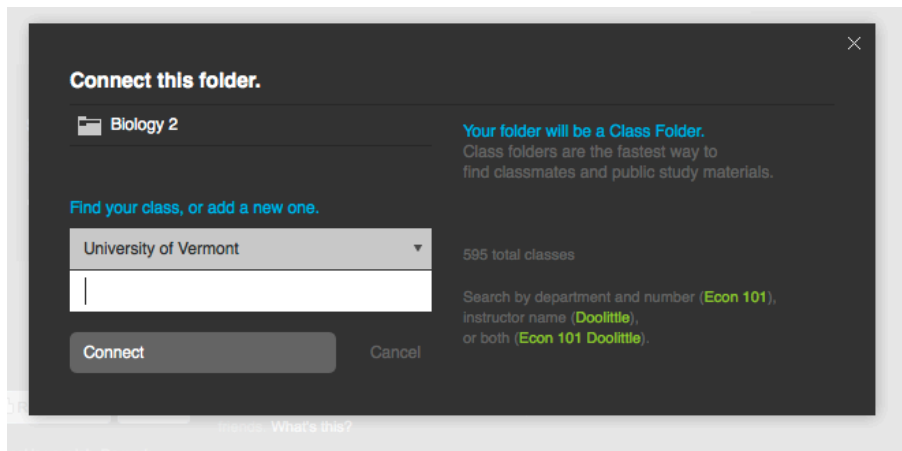
The screenshot shows the 'Manage Folders' page in the StudyBlue app. At the top, there's a navigation bar with the StudyBlue logo, a 'Backpack' dropdown, a user dropdown set to 'Rachel', an 'Inbox' icon, a 'New' button, and a search bar. Below the navigation bar, the user's name 'RACHEL CONRAD' is displayed. The main heading is 'Manage Folders'. Underneath, it says 'In my backpack' and '3 Folders'. A table lists the folders: 'EDSS 055 C' (8 items), 'GRE Vocab' (1 item), and 'Social, Historical, and Political Foundations' (0 items). Each folder has 'Archive.' and 'Disconnect.' links. At the bottom of the table, there's a text input field containing 'CHEM 31 Recitation' and 'Save' and 'Cancel' buttons. Below the table, there are 'Recommend' and 'Send' buttons, followed by a Facebook share link stating '20,901 people recommend this. Be the first of your friends. What's this?'. At the very bottom, there's a footer with the text 'Studying need not suck.', links for 'About / Mobile / FAQ / Legal / Blog / Facebook / Twitter', and '© 2011 STUDYBLUE Inc.'.

Step Three: Connect with other students in the class.

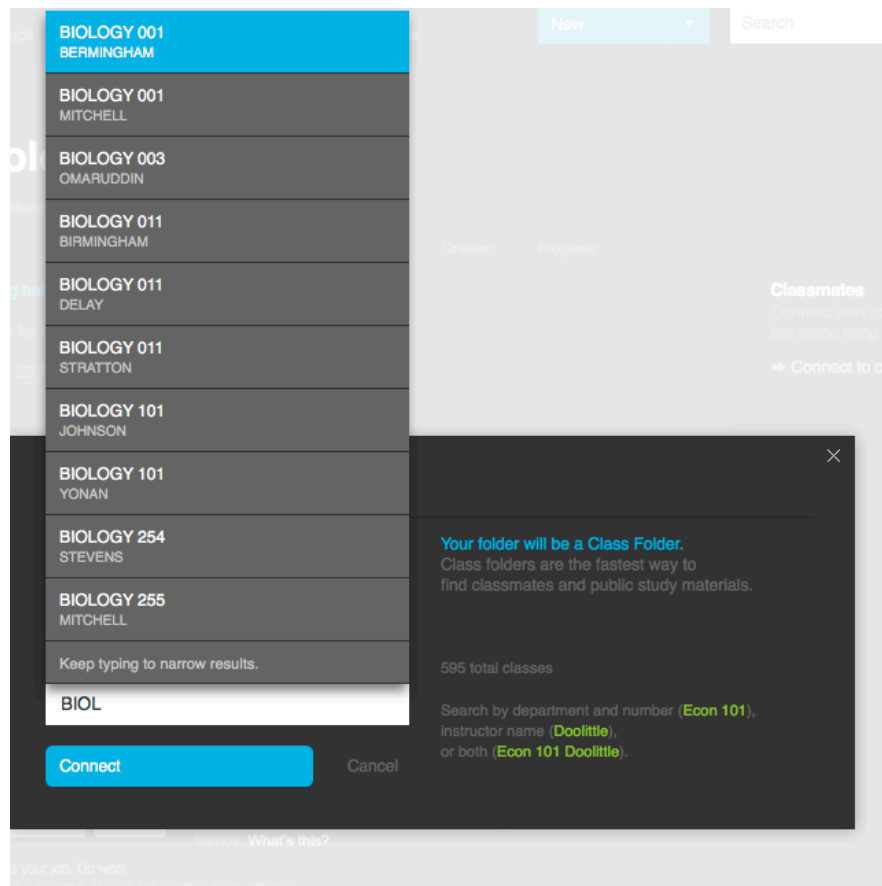
Courses are organized based on the University you attend, subject, course number, and Professor. Look to see if a classmate has already created a page for your class. Choose the “Connect to my Class” option on the left side of your screen.

The screenshot shows the 'Biology 2' class page in the StudyBlue app. The navigation bar at the top is identical to the previous screenshot. The main heading is 'Biology 2'. Below it, the text 'Study Materials' is displayed. There's a table with columns: 'Folder', 'Title', 'Size', 'Created', and 'Progress'. The 'Folder' column has a link 'Connect to my class.' The 'Title' column has the text 'Nothing here yet.' and a subtext 'Search for flashcards to study, or make your own.' At the bottom of the table, there are icons for adding a folder and a flashcard.

Choose the University of Vermont as your school.

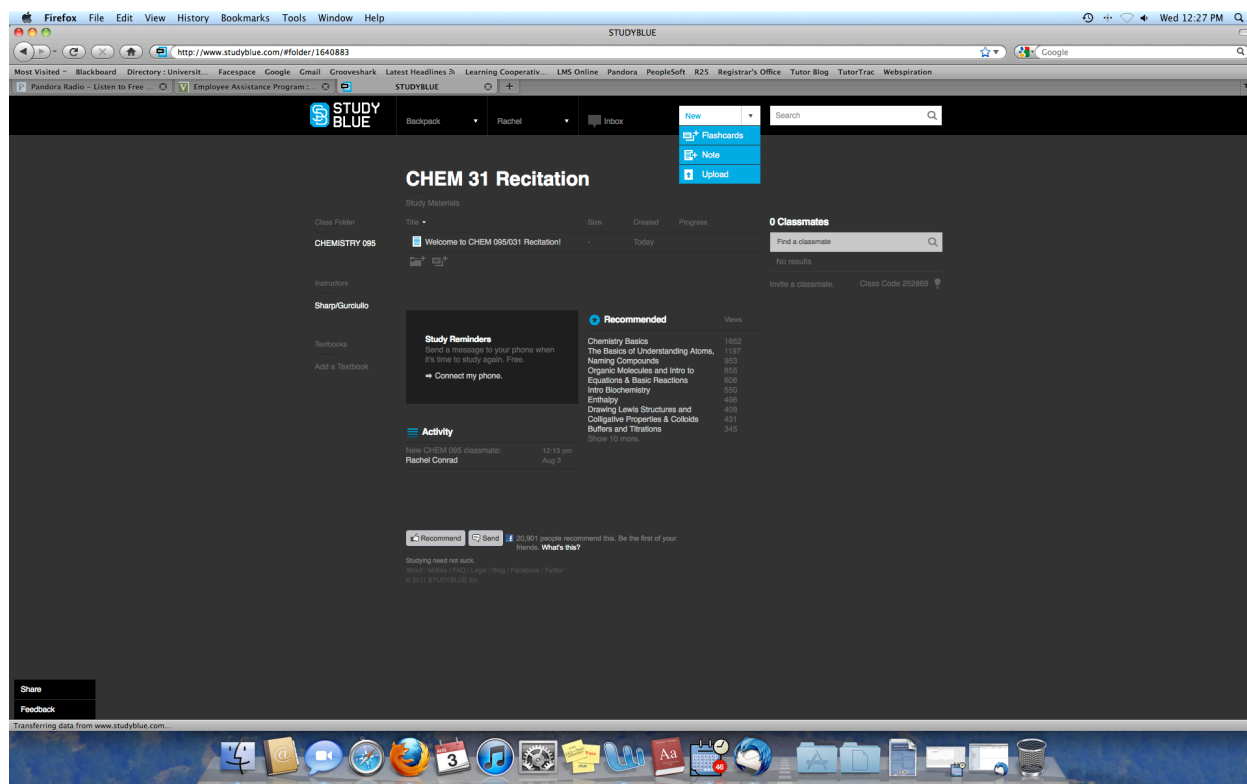


Type in your course code, and choose your course from the drop-down list. If you don't see a match, you can be the first to add and create the course.

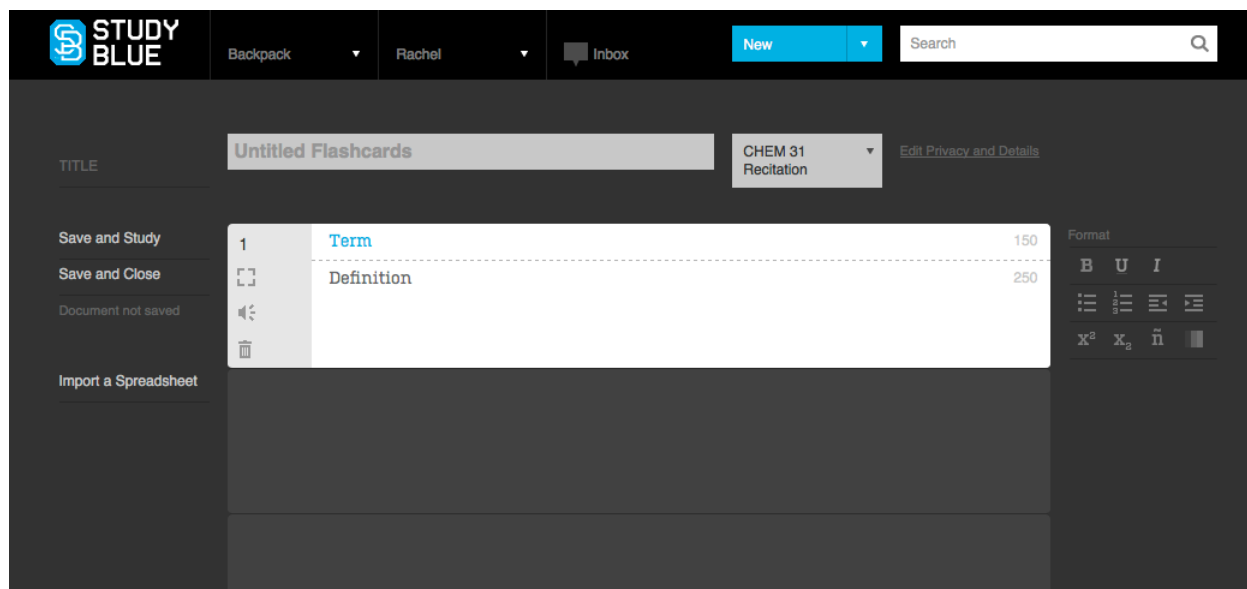


Step Four: Add new study tools to your course folder.

You have the option of uploading a document to share, adding a new note (which is like a blank document), and creating flashcards. From your course folder page, toggle over the “New” tab in the upper right side of the screen and choose which of the options you would like to create.



Add the information to custom flashcards or whichever tool you choose. Be sure to click “Save and Close” on the left of the screen to save your notes and flashcards for a later time.



Step Five: Use your notes cards and the resources other users have shared to study.

Things you have added will show up under the “Study Materials” heading. Also explore the shared resources under the “Recommended” heading, and flashcards or documents added by your classmates.

CHEM 31 Recitation

Study Materials

Class Folder	Title	Size	Created	Progress
CHEMISTRY 095	Welcome to CHEM 095/031 Recitation!	1 page	Today	

0 Classmates

Find a classmate

No results

Invite a classmate. Class Code 252869

Study Reminders

Send a message to your phone when it's time to study again. Free.

➔ Connect my phone.

Recommended

Topic	Views
Chemistry Basics	1656
The Basics of Understanding Atoms,	1197
Naming Compounds	955
Organic Molecules and Intro to	857
Equations & Basic Reactions	606
Intro Biochemistry	552
Enthalpy	496
Drawing Lewis Structures and	409
Colligative Properties & Colloids	431
Buffers and Titrations	345

Show 10 more.

Activity

New CHEM 095 classmate: Rachel Conrad 12:13 pm Aug 3

Once you have chosen a topic to study, choose “Study Now” on the top right side of the screen.

Chemistry Basics

No stats yet. Study now to begin tracking your progress.

Go Back

Flashcards Shuffled Study Now

Set a Study Reminder Study Later

Term	Definition	History
A gas is held in a container which h	1cm = 10mm	
Explain how you could exploit phys	A physical property of iron is that it	
a. Does electrolysis of water induce	a. Electrolysis of water induces a ch	
Why is it important that air is a mixt	The elements that comprise a comp	
Add the following values and use th		
Name an element that exists in gas	The three states of matter are solid	

Actions

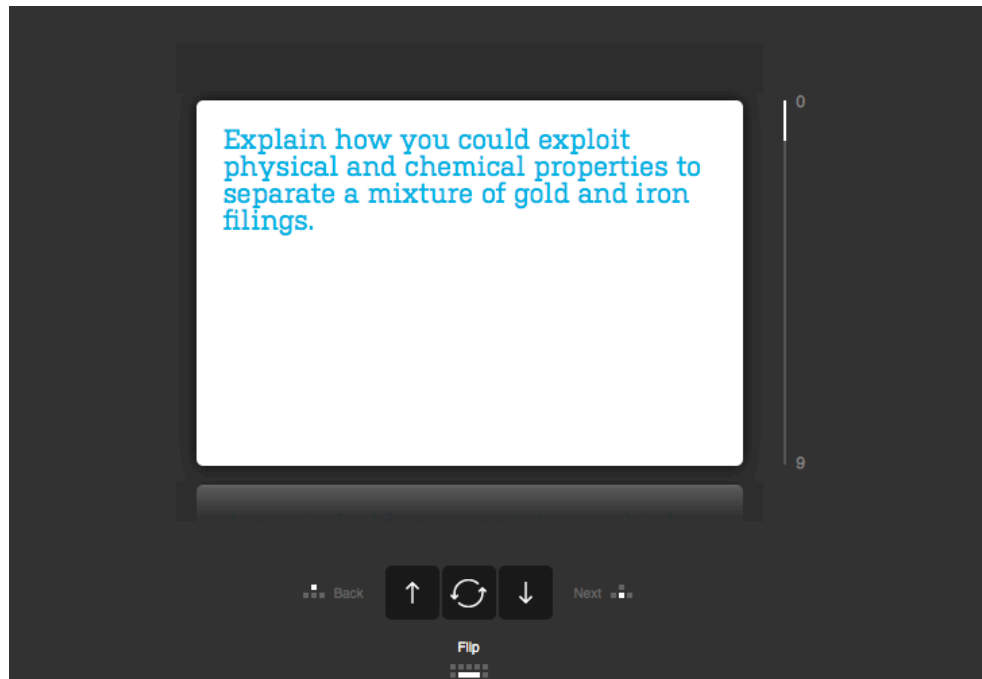
Save This

Share

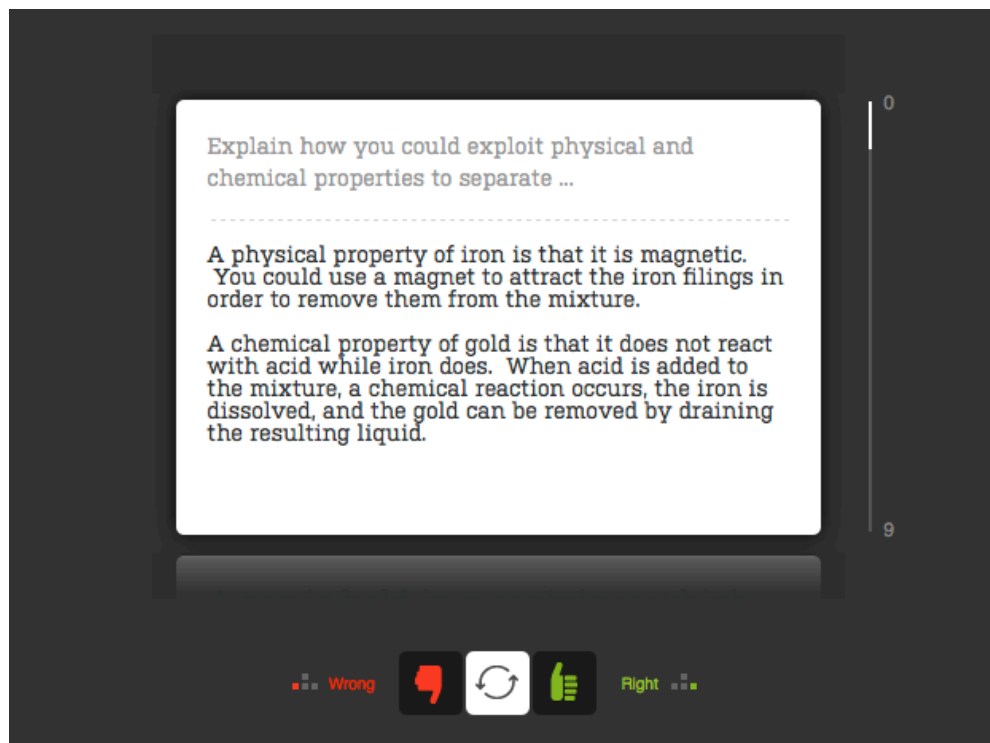
Like

Print


Try to answer the questions you see on the flashcard that appears. You can use the buttons on the bottom of the screen to move forward or backwards in your stack and to flip the card and see the answer.



Once you have flipped the card to check the answer, you can indicate whether or not you got the questions right. StudyBlue will keep track, and give you a grade when you're done.



The next time you study, you have the option of studying only those concepts you got incorrect the first time by choosing “Study the Wrongs.”

 Go Back

Chemistry Basics

[Study Stats™](#) [About Study Stats™](#)

Score

77%

9 Studied
7 Right
2 Wrong
0 Not Studied

Progress

100%
0%

77%
Estimated Mastery

Study Session No. 1

AllFlashcardsStudy the WrongsStudy Now

Term

Definition

History

1

Name an element that exists in each state of matter.

The three states of matter are solid, liquid, and gas.

4

Explain how you could exploit physical properties of iron.

A physical property of iron is that it is magnetic.

Hide. I know this. [BETTER](#)

2 cards filtered from 9.
[Clear Filters](#)

Set a Study Reminder

[Study Later](#)

Actions


[Save This](#)

Share

Like

Print

You can also choose to study the material on your flashcards in another format. Choose which format you'd like see the material in by choosing “Flashcards,” “Quiz,” or “Review Sheet” from the drop-down menu.

 Go Back

The Basics of Understanding Atoms, Ion and Molecules

No stats yet.
Study now to begin tracking your progress.

AllQuizFlashcardsQuizReview Sheet

ShuffledStudy Now

Definition

History

6

a. Nitrate ion

3

1. H₂ or O₂

2

Protons have a + (a) positive charge.

7

1

Name the three pieces of information needed to write an ionic formula.

a. atomic numberthe number of protons in the nucleus
b. mass numberthe number of protons and neutrons in the nucleus
c. Fe²⁺ is ferrous iron

5

4

a. negatively

All 7 cards.
[Clear Filters](#)

Set a Study Reminder

[Study Later](#)

Actions

[Save This](#)

Share

Like

Print

Info

Author
StudyBlue

Updated
07-04-11

Size
7 cards

Views
1196


Likes
4


0 Comments

Comments

No comments.
[Add a comment.](#)

StudyBlue will generate a multiple choice, fill in the blank, and short answer quiz for you, and grade it!



Options 

I'm done.

Multiple Choice5

1. Protons have a __ (a) ____ charge, while __ (b) ____ have a negative charge and neutrons have a __ (c) ____ charge. Atoms without a charge have an __ (d) ____ number of electrons and protons.

Fe2+ is ferrous iron

Fe3+ is ferric iron

Fe3+ is more positively charged

a. negatively

b. anion

c. Cations

d. positive

1. H₂ or O₂

2. H₂O, H₂O₂, CO, CO₂, CH₄, C₂H₄

3. NaCl, CaCl₂

a. positive

b. electron

c. neutral

d. equal

Special Characters
¶